

Name _____

Ions and Formulas Test Practice

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| 1. Chromic | 32. Sodium Hydroxide | 64. Nitrate |
| 2. Fluoride | 33. Neon | 65. Silver |
| 3. Sodium | 34. Water | 66. Sulfite |
| 4. Mercury (II) | 35. Chromium (III) | 67. Carbon Dioxide |
| 5. Calcium | 36. Barium | 68. Ferric |
| 6. Chloride | 37. Nitric Acid | 69. Carbonate |
| 7. Ammonia | 38. Chlorite | 70. Potassium |
| 8. Iodide | 39. Carbon Monoxide | 71. Carbon tetrachloride |
| 9. Sulfide | 40. Magnesium | 72. Mercury (I) |
| 10. Hydrogen Sulfate | 41. Chromate | 73. Hypochlorite |
| 11. Chlorate | 42. Dichromate | 74. Lead (IV) |
| 12. Copper (II) | 43. Aluminum | 75. Sulfate |
| 13. Hydrogen | 44. Iron (III) | 76. Permanganate |
| 14. Peroxide | 45. Hydrogen ion | 77. Nickelic |
| 15. Hydroxide | 46. Methane | 78. Lead (II) |
| 16. Hydrogen Carbonate | 47. Cuprous | 79. Proton |
| 17. Ferrous | 48. Ammonium | 80. Sucrose |
| 18. Chlorine | 49. Stannous | 81. Chromium (II) |
| 19. Tin (IV) | 50. Nitrite | 82. Sulfuric Acid |
| 20. Perchlorate | 51. Hydronium | 83. Hydrochloric Acid |
| 21. Beryllium | 52. Nitrogen | 84. Fluorine |
| 22. Zinc | 53. Bromine | 85. Oxygen |
| 23. Copper (I) | 54. Nickelous | 86. Iodine |
| 24. Iron (II) | 55. Mercurous | 87. Lithium |
| 25. Plumbous | 56. Plumbic | 88. Tin (II) |
| 26. Stannic | 57. Acetate | 89. Argon |
| 27. Sodium Chloride | 58. Bromide | 90. Cyanide |
| 28. Nickel (III) | 59. Diarsenic pentoxide | 91. Thiocyanate |
| 29. Nickel (II) | 60. Oxide | 92. Cupric |
| 30. Chromous | 61. He | 93. Mercuric |
| 31. Phosphate | 62. Glucose | 94. Oxalate |
| | 63. Thiosulfate | |